PRODUCT NAME	CAS#
Nitric Oxide	10102-43-9
TRADE NAME AND SYNONYMS	DOT I.D. NO.
Nitric Oxide, Nitrogen Oxide, Compressed	UN 1660
CHEMICAL NAME AND SYNONYMS	DOT HAZARD CLASS
Nitric Oxide	Division 2.3
	FORMULA
ISSUE DATE AND REVISIONS	NO
Revised July 2007	CHEMICAL FAMILY
Tevised bully 2007	Inorganic Gas

#### **HEALTH HAZARD DATA**

#### EMERGENCY OVERVIEW

Nitric Oxide is a colorless to reddish brown gas with suffocating odor. Poison gas. Corrosive. Oxidizer. Nitric Oxide is severely irritating to eyes and respiratory system. It is harmful if inhaled, respiratory tract irritation, skin irritation, blood damage. Containers may rupture or explode if exposed to heat. May ignite combustibles. May react on contact with water. May react on contact with air. Releases toxic, corrosive, flammable or explosive gases.

#### SYMPTOMS OF EXPOSURE

<u>Inhalation:</u> It will cause irritation, nausea, vomiting, stomach pain, chest pain, difficulty breathing, handsche, digginess, bluich skin color and lung congestion

headache, dizziness, bluish skin color and lung congestion.

<u>Skin Contact</u>: Irritation (possible severe) Eye Contact: Irritation (possibly severe)

#### TOXICOLOGICAL PROPERTIES

Toxicity Data: 1068 ng/m3/4 hour(s) inhalation-rat LC50

Local effects: Irritant: inhalation, skin. Acute Toxicity Level: Toxic: inhalation

Target Organs: blood

Medical conditions aggravated by exposure: respiratory disorders

# RECOMMENDED FIRST AID TREATMENT

PROMPT MEDICAL ATTENTION IS MANDATORY IN ALL CASES OF OVEREXPOSURE TO NITRIC OXIDE. RESCUE PERSONNEL SHOULD BE EQUIPPED WITH SELF-CONTAINED BREATHING APPARATUS.

<u>Inhalation</u>: If adverse effects occur, remove to uncontaminated area. Give artificial respiration if not breathing. Get immediate medical attention.

<u>Skin Contact:</u> Wash skin with soap and water for at least 15 minutes while removing contaminated clothing and shoes. Get medical attention, if needed. Thoroughly clean and dry contaminated clothing and shoes before reuse.

Eye Contact: Flush eyes with plenty of water for at least 15 minutes. Then get immediate medical attention.

Ingestion: If a large amount is swallowed, get medical attention.

Antidote: methylene blue, intravenous; ascorbic acid, intravenous.

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# HAZARDOUS MIXTURES OF OTHER LIQUIDS, SOLIDS, OR GASES

May ignite or explode on contact with combustible materials.

# **PHYSICAL DATA**

BOILING POINT	FREEZING POINT	
-242°F (-152 °C)	-263°F (-164°C)	
VAPOR PRESSURE @ 70°F	VAPOR DENSITY (AIR=1)	
26000 mmHg @ 20 °C	1.036	
SOLUBILITY IN WATER	VISCOSITY	
Very slightly (7.3% @ 0 °C)	0.0188 cP @ 25 °C	
EVAPORATION RATE	SPECIFIC GRAVITY (AIR=1)	
N/A	N/A	
APPEARANCE AND ODOR		
Colorless gas with suffocating odor		

# FIRE AND EXPLOSION HAZARD DATA

FLASH POINT (Method used)	AUTO IGNITION TEMPERATURE	FLAMMABLE LIMITS % BY VOLUME		
N/A	N/A	LEL N/	A UEL N/A	
EXTINGUISHING MEDIA		•	ELECTRICAL CLASSIFICATION	
	Water		Nonhazardous	
SPECIAL FIRE FIGHTING PROCEDURES				
Cut off the flow of gas which is supporting/accelerating fire if possible. Firefighters should wear				
respiratory protection and full turnout with additional chemical protective clothing as necessary to				
prevent exposure.				
UNUSUAL FIRE AND EXPLOSIO				
Nitric oxide is nonflammable but will support and may vigorously accelerate combustion.				

# **REACTIVITY DATA**

STABILITY		CONDITIONS TO AVOID		
Unstable				
Stable	X	N/A		
INCOMPATIBILITY (Materials to avoid)				
Oxidizing agents, halides, hydrocarbons and oxygen. Reacts vigorously with fluorine, fluorine				
oxides and chlorine in the presence of moisture.				
HAZARDOUS DECOMPOSITION PRODUCTS				
Oxidizes in air to form nitrogen dioxide, which is extremely reactive and a strong oxidizer. Upon				
contact with moisture and oxygen, it produces nitrous and nitric acids.				
HAZARDOUS POLYMERIZATION CONDITIONS TO AVOID				
May Occur				
Will Not Occur	X	N/A		

# **SPECIAL PROTECTION INFORMATION**

RESPIRTORY PROTECTION (Specify type)				
Positive pressure air line with full-face mask of	or self-contained breathing apparatus should be			
available for emergency use.				
VENTILATION	SPECIAL			
See Local Exhaust	N/A			
MECHANICAL (Gen.)	OTHER			
N/A	N/A			
LOCAL EXHAUST				
Local exhaust ventilation used in combination with partially or totally enclosed processes as				
necessary to control air contaminants to at or below acceptable exposure guidelines.				
PROTECTIVE GLOVES				
Any material	EYE PROTECTION			
	Gas-tight safety goggles or glasses			

#### OTHER PROTECTIVE EQUIPMENT

Safety shoes, safety shower and emergency eyewash station

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#### **SPILL OR LEAK PROCEDURES**

#### STEPS TO BE TAKEN IN CASE MATERIAL IS RELEASED OR SPILLED

Stop leak if possible without personal risk. Evacuate all personnel from affected area. Use appropriate protective equipment. If leak is in container or container valve, contact HSG for special advice.

#### WASTE DISPOSAL METHOD

Do not attempt to dispose of waste or unused quantities. Return in the shipping container properly labeled, with any valve outlet plugs or caps secured and valve protection cap in place to your supplier. For emergency disposal assistance, contact HSG for special advice.

DOT Hazard Class: Division 2.3

UN 1660

ID No.

## **SPECIAL PRECAUTIONS\***

### SPECIAL LABELING INFORMATION

DOT Shipping Name: Nitric Oxide, Compressed DOT Shipping Label: Poision Gas, Oxidizer, Corrosive

SPECIAL HANDLING RECOMMENDATIONS

Use only in well-ventilated areas. Valve protection caps must remain in place unless container is secured with valve outlet piped to use point. Do not drag, slide or roll cylinders. Use a suitable hand truck for cylinder movement. Use a pressure reducing regulator when connecting cylinder to lower pressure (<3,000 psig) piping or systems. Do not heat cylinder by any means to increase the

discharge rate of product from the cylinder. Use a check valve or trap in the discharge line to prevent

hazardous back flow into the cylinder.

Never carry a compressed gas cylinder or a container of a gas in cryogenic liquid form in an enclosed space such as a car trunk, van or station wagon. A leak can result in a fire, explosion, asphyxiation or a toxic exposure.

#### SPECIAL STORAGE RECOMMENDATIONS

Protect cylinders from physical damage. Store in cool, dry, well-ventilated area away from heavily trafficked areas and emergency exits. Do not allow the temperature where cylinders are stored to exceed 125°F. Cylinders should be stored upright and firmly secured to prevent falling or being knocked over. Full and empty cylinders should be segregated. Use a "first in - first out" inventory system to prevent full cylinders being stored for excessive periods of time.

### OTHER RECOMMENDATIONS OR PRECAUTIONS

Compressed gas cylinders should not be refilled except by qualified producers of compressed gases. Shipment of a compressed gas cylinder which has not been filled by the owner or with his (written) consent is a violation of Law. Always secure cylinders in an upright position before transporting them. NEVER transport cylinders in trunks of vehicles, enclosed vans, truck cabs or in passenger compartments. Transport cylinders secured in open flatbed or in open pick-up type vehicles.

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